DR. BABASAHEB AMBEDKAR TECHNOLOGICAL UNIVERSITY, LONERE Supplementary Summer - 2023

Course: B. Tech Sem: I

Subject: Engineering Chemistry

Subject code: BTBS102

Marks: 60 Date:09/08/2023 Duration: 3 Hr.

Instructions to the Students:

- 1. All the questions are compulsory.
- 2. Use of non-programmable scientific calculators is allowed.
- 3. Draw a neat diagram whenever necessary.
- 4. Figures to right indicate full marks.

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		CO)	
Q.1	Solve Any Two of the following.		
	A) What are the disadvantages of hard water in Domestic and Industrial use?	2	06
	B) Explain Ion exchange process of softening of water.C) Explain the determination of Dissolved Oxygen by Winkler's	1	06
	Method.	2	06
Q.2	Solve Any Two of the following.		
	A) Describe Phase diagram of two components Ag – Pb alloy system.	2	06
	B) Explain phase diagram of one component Water system.	2	06
	C) Discuss the term component and degrees of freedom involved in		
	Phase rule equation with examples.	1	06
Q.3	Solve Any Two of the following.		
	A) Write a note on Galvanic corrosion.	1	06
	B) Explain any one method to minimize the rate of corrosion.	2	06
	C) Describe in brief Direct chemical (Dry) corrosion.	2	06
Q.4	Solve Any Two of the following.		
	A) Explain Proximate analysis of coal.	2	06
	B) Write a note on: Refining of Petroleum.	2	06
	C) Describe Chemical properties of lubricant.	1	06
Q.5	Solve Any Two of the following.		
	A) Describe Conductometric titration with any two examples.	1	06
	B) Explain Ostwald's theory of Acid –Base indicators.	2	06
	C) Explain conductance measurement of a solution by Wheatstone's		
	bridge method.	2	06